S67E2W7

SUMMATIVE ASSESSMENT – I, 2015-16 Class - X

Time Allowed: 3 hours

Maximum Marks: 90

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General Instructions:

- 1. The question paper comprises of two Sections, A and B. You are to attempt both the sections.
- 2. All questions are compulsory
- 3. All questions of Section-A and all questions of Section-B are to be attempted separately.
- 4. Question numbers 1 to 3 in Section-A are one mark questions. These are to be answered in one word or in one sentence
- 5. Question numbers 4 to 6 in Sections-Aaretwo marks questions. These are to be answered in about 30 words each.
- 6. Question numbers 7 to 18 in Section-A are three marks questions. These are to be answered in about 50 words each
- 7. Question numbers 19 to 24 in Section-A are five marks questions. These are to be answered in about 70 words each.
- 8. Question numbers 25 to 33 in Section-B are multiple choice questions based on practical skills. Each question is a one mark question. You are to select one most appropriate response out of the four provided to you.
- 9. Question numbers 34 to 36 in Section-B are questions based on practical skills. Each question is of two marks.

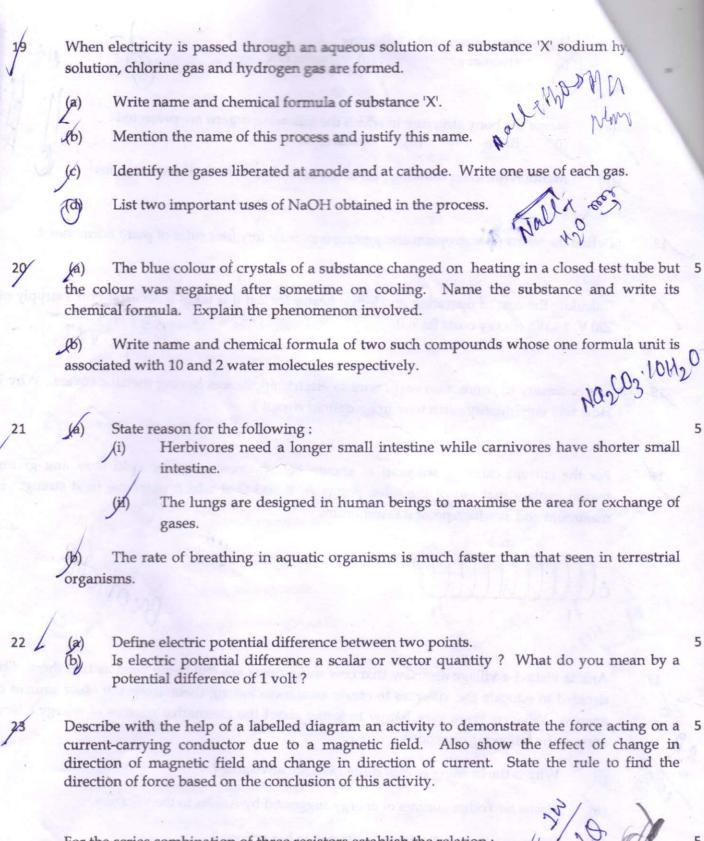
SECTION-A

- Name the process used by single-celled organisms for taking in food, exchange of gases or 1 removal of wastes. A compass needle shows deflection when brought near a current carrying conductor. Why? Mention the minimum temperature difference required between surface water and the water ' 1 at depths of up to 2 km in an ocean thermal energy plant.
- State what happens when zinc granules are heated with sodium hydroxide solution. Write 2 the balanced chemical equation for this reaction. Name the main product formed in this reaction.

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15	Consider the following chemical reaction:
	X + Barium chloride → Y + Sodium chloride
	(white ppt.)
	Identify: (a) X and Y; (b) The type of reaction.
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6	What is synapse? How does a message of an impulse transmit through a synapse?
2	Identify the type of each of the following reactions. Also write balanced chemical equation for each.
	(a) The reaction mixture becomes warm.
	(b) An insoluble substance is formed.
8	(a) What is the action of litmus on: (i) dry ammonia gas
	(iii) solution of ammonia gas in water
an hyarden	(h) State the observations you would make an adding sodium hydroxide to aqueous solution of :
physical fill	11 + Alla
· Lo	aluminium chloride.
Cost of	Give balanced chemical equations.
9	Give balanced chemical equations. Explain the following terms: (a) Ore (b) Mineral (c) Gangue AL203 AL2
	Explain the following terms: (a) Ore (b) Mineral (c) Gangue Sodium is kept immersed in kerosene oil. (b) Iron the most widely used metal is never used in its pure state.
10	Give reason for the following:
	(a) Sodium is kept immersed in kerosene oil.
11 11	Iron, the most widely used metal, is never used in its pure state
H 200 2	Hydrogen is not evolved when a metal reacts with nitric acid (except Mn and Mg)
and at he	man demonstration and small and second and making and manager bounded Character and All All Andrews Character and Andrews Character and All Andrews Character and An
11	(a) State reason for the following: (ii) Rings of cartilage are present in the trachea. (iii) Plants look green in colour. Nasoy+ BaCl ₂ — Rasoy+ LNacl. + Nasoy (iii) Plants look green in colour.
	Nasoy+ Bacl2 -> Bason + I Nacl. + No
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	(b) Write other names of the following: (i) alveolar sac (ii) voice box
12	(a) Name the bony structure in which the following organs are protected: (i) Brain (ii) Spinal cord
	(b) Mention the main thinking part of the brain and write its three functions.
	Define the terms phototropism and geotropism. State any four roles of plant hormones.
14	Calculate the cost of operating an electric heater for 2 h if it takes 5 A current on a supply of 220 V, 1 kWh energy costs Rs. 6.0.
15	It is necessary to connect an earth wire to electric appliances having metallic covers. Why? How will you identify earth wire in household circuit?
18	For the current carrying solenoid as shown below, draw magnetic field lines and giving reason explain that out of the three points A, B and C at which point the field strength is maximum and at which point it is minimum.
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¹⁷	Ankita visited a village and saw that cow dung cakes are still being used as fuel there. She decided to educate the villagers to create awareness among them about the other sources of energy. Villagers were very happy to know about the alternative sources of energy. Now answer the following questions:
00	(i) Why is the burning of cow dung cake not advisable?
	(iii) Name two other sources of energy suggested by Ankita to the villagers.
	(jii) What qualities of Ankita are reflected in her actions?
18	Energy can neither be created nor be destroyed. In the context of the statement explain, why do we talk about energy crisis?
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For the series combination of three resistors establish the relation:

 $R = R_1 + R_2 + R_3$

where the symbols have their usual meanings.

Calculate the equivalent resistance combination of three of the resistors 6Ω , 9Ω and 18Ω joined in parallel.





25	Shruti placed a drop of solution 'A' on a strip of pH paper and a deep blue colour was produced.	The	1
	solution 'A' is of :		

- dil NaOH solution (a)
- dil hydrochloric acid (b)

(c) water (d) dil ethanoic acid

26 During the performance of an experiment to test the pH of given samples by using pH paper four 1 students I, II, III and IV recorded the following observations:

Student	Sample taken	pH paper turns to
1 4 / / >	Dil hydrochloric acid	Yellow
11	Water	Green
III	Dil sodium bicarbonate solution	Light blue
IV	Dilsodium hydroxide solution	Violet

Which one of the above observations is incorrect?

- (II)
- (c)

(111)

- (d)
- (IV)

While performing the experiment to 'Study the properties of HCl and NaOH by their reaction 1 27 with Na2CO3 a student can confirm whether hydrogen or carbon dioxide is liberated, by :

- their colour (a)
- their odour (b)
- lime water test (c)
- (d) pH paper

Tanisha put iron filing in different test tubes containing ZnSO₄, Al₂(SO₄)₃, FeSO₄ and CuSO₄. The one in which she observes a change in colour will be the one which contains:

(a)

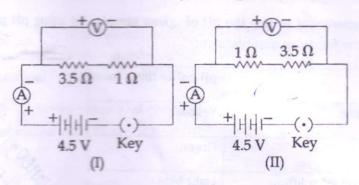
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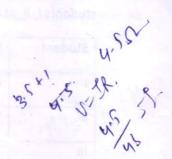
- Al₂(SO₄)₃ FeSO₄ (c) ZnSO₄

CuSO₄

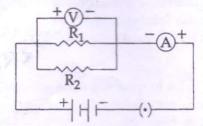
29 Disha took two iron nails and put them in aluminium sulphate solution. After sometimes she 1 observed that:

- (a) the solution becomes warm
- (b) grey-metal is deposited on the iron nail
- (c) the colourless solution changes to light green
- (d) solution remains colourless and no deposition is observed on the iron nail.
- To determine the equivalent resistance of two resistors connected in series, a student prepared two electric circuits, correct reading of ammeter in the circuits is:





- (a) In circuit I, 1.0 A and in II, 0.1 A
- (b) In both circuits I and II, 1.0 A
- (c) In circuit I, 0.1 A and in II, 1.0 A
- (d) In both circuits I and II, 0.1 A
- To calculate the equivalent resistance when R₁ and R₂ are connected in parallel, a student is 1 not able to perform the experiment for given circuit diagram.



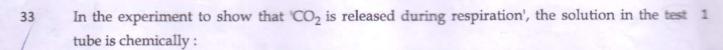
The correct reason is:

- (a) Position of Voltmeter is not correct.
- (b) Position of Ammeter is not correct.
- (c) Terminals of Voltmeter are not connected correctly.
- (d) Terminals of Ammeter are not connected correctly.

In an experiment on photosynthesis, students were instructed to cover a portion of a leaf of a 1 de-starched potted plant with an opaque paper.

"A" covered one of the leaves with red strip, "B" with green, "C" with blue and "D" with black. When the starch test was done on the leaves after 4 hours, the result showed no starch in:

- (a) The portion covered with red, green and blue strips
- (b) The portion covered with green strip
- (c) The portion covered with black and blue strips
- (d) Any of the covered portions.



- (a) NaOH
- (b) KOH (c)
- NaCl
- (d) KCl
- Write two precautions that must be taken while studying the decomposition reaction, when 2 the ferrous sulphate crystals are heated in a boiling tube.
- Draw a diagram of a circuit showing a resistor and a voltmeter connected in parallel.
- Explain why leaf is preferred for the preparation of temporary mount to show stomata. State 2 two functions of stomata.

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