## JEE Main 2023 (Memory based)

## 24 January 2023 - Shift 2

Answer \& Solutions

## PHYSICS

1. The corresponding gate for the given circuit diagram is
A. OR
B. AND
C. NAND
D. XOR

## Answer (C)

Sol.


When both $A$ and $B$ are open or any one of them is open, bulb will glow. When both switches are closed, due to short circuit, bulb will not glow.

| A | B | Output |
| :--- | :--- | :--- |
| 0 | 0 | 1 |
| 0 | 1 | 1 |
| 1 | 0 | 1 |
| 1 | 1 | 0 |

Output $=\overline{A . B}$
2. A rod ' $O A^{\prime}$ ' of length ' $L$ ' is rotating with constant angular speed ' $\omega$ ' about an axis passing through point $O$ in a perpendicular uniform magnetic field ' $B$ '. Find the emf induced between points $O$ and $A$ of the rod.
A. $\frac{1}{2} B \omega L^{2}$
B. $B \omega L^{2}$
C. $\frac{3}{2} B \omega L^{2}$
D. $2 B \omega L^{2}$


## Answer (A)

## Sol.

$$
\begin{aligned}
& d \varepsilon=B v d x=B \omega x d x \\
& \int d \varepsilon=\int_{0}^{L} B \omega x d x \\
& \varepsilon=\frac{B \omega L^{2}}{2}
\end{aligned}
$$


3. Let $\gamma_{1}$ be the ratio of molar specific heat at constant pressure and constant volume for a monoatomic gas and let $\gamma_{2}$ be this ratio for a diatomic gas. Find $\frac{\gamma_{1}}{\gamma_{2}}$
A. $21 / 25$
B. $7 / 3$
C. $25 / 21$
D. $3 / 7$

## Answer (C)

## Sol.

We know that $\gamma_{1}=\frac{5}{3}$ and $\gamma_{2}=\frac{7}{5}$

$$
\frac{\gamma_{1}}{\gamma_{2}}=\frac{25}{21}
$$

4. A right-angled triangle has current of 2 A . The edge length are shown in the diagram. Magnetic field is acting in the plane of the triangle. The magnetic force acting on wire $A B$ is
A. $\frac{5}{130} N$
B. $\frac{15}{2} N$
C. $\frac{3}{40} N$
D. $\frac{9}{130} \mathrm{~N}$

Answer (D)

Sol.

$$
\begin{aligned}
& B=0.75 \mathrm{~T} \\
& \left|\vec{F}_{A B}\right|=i \times\left(l_{A B}\right) B \sin \theta \\
& =2 \times\left(\frac{5}{100}\right) \times 0.75 \times \frac{12}{13} \\
& =\frac{9}{130} N
\end{aligned}
$$

5. Assertion $(A)$ : Steel is used to build big structures.

Reason $(R)$ : Steel has more elastic modulus as compared to other materials.
A. Both $A$ and $R$ are true and $R$ is the correct explanation of $A$.
B. Both $A$ and $R$ are true but $R$ is not the correct explanation of $A$.
C. $A$ is true but $R$ is false.
D. Both $A$ and $R$ are false.

## Answer (A)

## Sol.

Large elastic modulus implies deformations will stay in elastic region and permanent deformations will not occur for small magnitude of force.
6. If $\vec{a}=\hat{\imath}+2 \hat{\jmath}+m \hat{k}$ and $\vec{b}=\hat{\imath}-2 \hat{\jmath}+m \hat{k}$ are perpendicular to each other, then $m=\sqrt{x}$. Find $x$.

## Answer (3)

## Sol.

$$
\begin{aligned}
& \vec{a}=\hat{\imath}+2 \hat{\jmath}+m \hat{k} \text { and } \vec{b}=\hat{\imath}-2 \hat{\jmath}+m \hat{k} \\
& \vec{a} \perp \vec{b} \Rightarrow \vec{a} \cdot \vec{b}=0 \\
& 1-4+m^{2}=0 \\
& m=\sqrt{3} \\
& x=3.00
\end{aligned}
$$

7. The distance of earth from the sun is $1.5 \times 10^{8} \mathrm{~km}$. If the time period of an imaginary planet is 2.83 years, then its distance from the sun comes out to be $n \times 10^{8} \mathrm{~km}$. Find $n$.

## Answer (3)

## Sol.

$$
\begin{aligned}
& T^{2} \propto a^{3} \\
& {\left[\frac{T^{2}}{a^{3}}\right]_{\text {earth }}=\left[\frac{T^{2}}{a^{3}}\right]_{\text {any planet }}} \\
& \frac{1}{\left(1.5 \times 10^{8}\right)^{3}}=\frac{2.83^{2}}{(x)^{3}} \\
& \begin{array}{l}
x=3 \times 10^{8} \\
n=3
\end{array}
\end{aligned}
$$

8. For the given electrical circuit, the switch ' $s$ ' is closed at $t=0$. Find the current (in $A$ ) drawn through the battery at steady state.


## Answer (3)

## Sol.

At steady state, inductors behave as shorted.

9. The velocity-time $(v-t)$ graph for a particle moving along a straight path is shown. Find the time taken (in seconds) by the particle to cover the first 10 m of distance.


Answer (5)
Sol.
Distance $=$ Area under $v-t$ graph
Area in first 5 seconds,
$A=\frac{1}{2} \times t \times 4=10 \mathrm{~m}$
$t=5 s$
10. An electron jumps from $3^{r d}$ excited state to the ground state in a hydrogen atom. The wavelength of photon emitted in nm to the nearest integer is $\qquad$ ( $h c=1240 \mathrm{eV}-\mathrm{nm}$ )

## Answer (97)

Sol.

$$
\begin{aligned}
& n_{i}=4, n_{f}=1 \\
& \Delta E=13.6\left(1-\frac{1}{16}\right)=\frac{13.6 \times 15}{16}=12.75 \mathrm{eV} \\
& \lambda(\mathrm{~nm})=\frac{1240}{12.75} \approx 97.25 \mathrm{~nm} \\
& \lambda(\mathrm{~nm})=97 \mathrm{~nm}
\end{aligned}
$$

11. The velocity-Time graph of a body moving along straight line is given as shown. The ratio of displacement and distance is:
A. $1: 1$
B. $1: 2$
C. $1: 3$
D. $1: 4$

Answer (C)


Sol.
Distance $=8 \times 2+4 \times 2+4 \times 4+4 \times 2=48 \mathrm{~m}$
Displacement $=8 \times 2-4 \times 2+4 \times 4-4 \times 2=16 \mathrm{~m}$
Ratio $=1: 3$
12. A copper wire is elongated such that its length is increased by $20 \%$. Then the percentage increase in the resistance is
A. $20 \%$
B. $30 \%$
C. $44 \%$
D. $50 \%$

Answer (C)

## Sol.



Before Heating

$$
R_{o}=\left(\frac{\rho l_{o}}{A_{o}}\right)
$$

After Heating

Volume $=$ Constant

$$
\begin{aligned}
& A_{o} l_{o}=1.2 l_{o} A \\
& A=\left(\frac{A_{o}}{1.2}\right)
\end{aligned}
$$

So,
$R=\frac{\rho l}{A}=\frac{\rho \times\left(1.2 l_{o}\right)}{\left(\frac{A_{o}}{1.2}\right)}=\frac{\rho l_{o}}{A_{o}} \times 1.44=1.44 R_{o}$
Percentage increase in $=\frac{R-R_{o}}{R_{o}} \times 100=44 \%$
13. Assertion (A): Acceleration due to gravity decreases with both height and depth from earth surface.

Reason (R): If height and depth are equal for two points from surface of the earth, then acceleration due to gravity will be same at those points.
A. Both $A$ and $R$ are true and $R$ is the correct explanation of $A$
B. Both $A$ and $R$ are true and $R$ is not the correct explanation of $A$
C. $A$ is true but $R$ is false.
D. Both $A$ and $R$ are false

## Answer (C)

## Sol.

Acceleration due to gravity is maximum on the surface.
It increases linearly from centre to surface and then decreases by inverse square law.
So, Assertion is true, and Reason is false.
14. Two concentric semi-circular rings (radii $R_{1}$ and $R_{2}$ ) have equal linear charge density ( $\lambda$ each) as shown:

Find the potential at center $C$.
A. $\frac{\lambda}{\epsilon_{0}}$
B. $\frac{2 \lambda}{\epsilon_{0}}$
C. $\frac{\lambda}{4 \epsilon_{0}}$
D. $\lambda / 2 \epsilon_{0}$


## Answer (D)

## Sol.



## At point C:

$$
\begin{aligned}
& V_{c}=V_{1}+V_{2} \\
& V_{c}=\frac{1}{4 \pi \epsilon_{0}} \times \frac{\lambda_{1}\left(\pi R_{1}\right)}{R_{1}}+\frac{1}{4 \pi \epsilon_{0}} \times \frac{\lambda_{1}\left(\pi R_{2}\right)}{R_{2}}=\frac{\lambda_{1}(\text { or } \lambda)}{2 \epsilon_{0}}
\end{aligned}
$$

15. Electric field vector and magnetic field vector of an electromagnetic wave is given as $\vec{E}=E_{0} \sin (\omega t-k x) \hat{\jmath}$ and $\vec{B}=B_{0} \sin (\omega t-k x) \widehat{k}$, then choose the correct option.
A. $\omega E_{0}=k B_{0}$
B. $k E_{0}=\omega B_{0}$
C. $k \omega=E_{0} B_{0}$
D. None of these

## Answer (B)

## Sol.

$$
\begin{aligned}
& \text { Given: } \vec{E}=E_{0} \sin (\omega t-k x) \hat{\jmath} \text { and } \vec{B}=B_{0} \sin (\omega t-k x) \hat{k} \\
& \begin{array}{c}
E_{0} \\
v \\
v \\
=\frac{\omega}{k}
\end{array} \\
& E_{0}=\frac{\omega}{k} B_{0} \Rightarrow k E_{0}=\omega B_{0}
\end{aligned}
$$

16. Consider the circuit shown:

Resistance of each voltmeter is $400 \Omega$. Find the reading of any one voltmeter.

A. 60 V
B. 45 V
C. 80 V
D. 30 V

## Answer (B)

Sol.
By symmetry, voltage would be divided equally.
$\Rightarrow$ Voltage across a voltmeter $=\frac{90}{2} \mathrm{~V}=45 \mathrm{~V}$
17. A long solenoid has 70 turns per cm and carries current 2 A . The magnetic field inside the solenoid is $\qquad$ ( $\mu_{o}=4 \pi \times 10^{-7}$ in SI units)
A. $125.2 \times 10^{-4} \mathrm{~T}$
B. $835.2 \times 10^{-4} \mathrm{~T}$
C. $176.0 \times 10^{-4} \mathrm{~T}$
D. $880 \times 10^{-4} \mathrm{~T}$

Sol.
Magnetic field inside solenoid $=\mu_{o} n i$
where $n=$ Number of turns per unit length

$$
\begin{aligned}
& n=7000 \text { turns } / \mathrm{m} \\
& \begin{aligned}
B_{\text {solenoid }} & =\mu_{o} n i=\left(4 \pi \times 10^{-7} \times 7000 \times 2\right) T \\
& =8 \times \frac{22}{7} \times 10^{-7} \times 7000 \mathrm{~T} \\
& =176 \times 10^{-4} \mathrm{~T}
\end{aligned}
\end{aligned}
$$

18. A ray of parallel white light incidence on convex lens and colour gets separated, this phenomenon is called
A. Chromatic aberration
B. Polarization
C. Scattering
D. Spherical aberration

## Answer (A)

Sol.
This phenomenon is known as chromatic aberration.

19. Assertion (A): The weight of an object at Mount Everest is lesser than its weight at sea level.

Reason (R): The value of $g$ decreases as height increases.
A. Assertion and Reason both are correct, and Reason is the correct explanation of Assertion.
B. Assertion and Reason both are correct, but Reason is not the correct explanation.
C. Assertion is correct reason is incorrect.
D. Assertion is incorrect reason is correct.

## Answer (A)

## Sol.

As we know that:

$$
g=\frac{G M}{(R+h)^{2}}, \text { Weight }=m g
$$

As height from surface of earth increases $g$ decreases, so weight on Mount Everest is less than the weight on the surface of earth.
20. A convex lens of refractive index 1.5 has focal length 20 cm in air. When this lens is immersed in water (refractive index $=4 / 3$ ), the new focal length (in cm) is $\qquad$ -.

## Answer (80)

Sol.

$$
f_{\text {air }}=20 \mathrm{~cm}, \mu_{\text {water }}=\frac{4}{3}, \mu=1.5
$$

When convex lens is in air,

$$
\begin{equation*}
\frac{1}{f_{\text {air }}}=(\mu-1)\left(\frac{2}{R}\right)=0.5\left(\frac{2}{R}\right) \tag{1}
\end{equation*}
$$

When lens is immersed in water,

$$
\begin{equation*}
\frac{1}{f}=\left(\frac{\mu}{\mu_{\text {water }}}-1\right)\left(\frac{2}{R}\right)=0.125\left(\frac{2}{R}\right) \tag{2}
\end{equation*}
$$

From (1) and (2),

$$
\begin{aligned}
& \frac{f}{f_{\text {air }}}=\frac{0.5}{0.125} \\
& f=0.5 \times \frac{20}{0.125}=80 \mathrm{~cm}
\end{aligned}
$$

## CHEMISTRY

1. In which of the following metal extraction both Oxidation and Reduction process are involved?
A. Au
B. Cu
C. Fe
D. Al

## Answer (A)

## Sol.

In the extraction of Gold (Au) first oxidation of gold takes place and then it is reduced into gold.
At Anode, $\quad 4 \mathrm{Au}(s)+8 \mathrm{CN}^{-}(a q)+2 \mathrm{H}_{2} \mathrm{O}+\mathrm{O}_{2}(\mathrm{~g}) \rightarrow 4\left[\mathrm{Au}(\mathrm{CN})_{2}\right]^{-}(a q)+40 \mathrm{H}^{-}(a q)$
At Cathode, $2\left[M(C N)_{2}(a q)+Z n(S) \rightarrow\left[Z n(C N)_{2}\right]^{2-}(a q)+2 A u(s)\right.$
2. $\alpha$ - particle, proton and electron have same kinetic energy. Select the correct order of de Broglie wavelength
A. $\lambda_{p}=\lambda_{\alpha}=\lambda_{e}$
B. $\lambda_{e}>\lambda_{p}>\lambda_{\alpha}$
C. $\lambda_{\alpha}>\lambda_{e}>\lambda_{p}$
D. $\lambda_{p}>\lambda_{e}=\lambda_{\alpha}$

## Answer (B)

## Sol.

As we know,
$\lambda=\frac{h}{\sqrt{2 m(K \cdot E)}}$
$\lambda=$ De Broglie wavelength
$m=$ Mass of the Particles, $k E=$ Kinetic Energy of the Particle
$h=$ Planck's Constant $=6.6 \times 10^{-34}$ Joule Second
Mass of $\alpha$ - particle $>$ mass of proton $>$ mass of electron
$\lambda$ is inversely proportional to the mass of the particle.
So, higher the mass of the particle less will be the de Broglie wavelength associated with it so option (B) is the correct option.
3. Find out order of reaction of decomposition of $A B_{3}(g)$ using the given information

| Initial pressure $(\mathrm{mmHg})$ of $A B_{3}(g)$ | 50 | 100 | 200 | 400 |
| :--- | :--- | :--- | :--- | :---: |
| $t_{\frac{1}{2}}(\mathrm{sec})$ | 4 | 2 | 1 | 5 |

A. 0
B. 1
C. 2
D. 3

## Answer (C)

Sol.
As we know,

$$
\left(t_{\frac{1}{2}}\right) \alpha\left(P_{0}\right)^{1-n}
$$

By observation,

$$
\frac{4}{2}=\left(\frac{50}{100}\right)^{1-n}
$$

$2=\left(\frac{1}{2}\right)^{1-n}$
$2^{1}=2^{(n-1)}$
$1=(n-1)$
So, $n=2$
second order reaction so correct option is (C).
4. Predict the hybridization state of the central metal ion and magnetic nature of the complex $\left[\mathrm{Co}\left(\mathrm{NH}_{3}\right)_{6}\right]^{3+}$.
A. $s p^{3} d^{2}$, Paramagnetic
B. $s p^{3} d^{2}$, Diamagnetic
C. $d^{2} s p^{3}$, Paramagnetic
D. $d^{2} s p^{3}$, Diamagnetic

## Answer (D)

Sol.
Oxidation state of cobalt in the given complex is +3

$$
C o^{3+}: 3 d^{6}
$$



$$
\left[\mathrm{Co}\left(\mathrm{NH}_{3}\right)_{6}\right]^{3+}
$$

As $\left(\mathrm{NH}_{3}\right)$ act as strong field ligand so pairing will be take place here,


Hybridisation state of $C o^{3+}: d^{2} s p^{3}$
Since no unpaired electron is present here so Magnetic nature is Diamagnetic and Correct option is (D)
5. Consider the following reaction


P is?
A.

B.

C.

D.


Answer (A)

Sol.

6. Which graph is correct for Isothermal process at $T_{1}, T_{2} \& T_{3}$ if $\left(T_{3}>T_{2}>T_{1}\right)$

A.
B.

C.

D.


## Answer (D)

Sol.
According to Boyle Law $P \propto \frac{1}{V}$
The graph must be hyperbola.
As we know, $P V=n R T$
So as increase the Temperature the PV graph area increases.


As $\left(V_{3}>V_{2}>V_{1}\right)$ for fixed P
$=\left(T_{3}>T_{2}>T_{1}\right)$
And the correct option is (D)
7. The number of Peptide bonds present in Tripeptide VAL - PRO - GLY is
A. 1
B. 2
C. 3
D. 4

## Answer (B)

Sol.
In between two Amino Acids there are 1 peptide bond in question 3 Amino acids are given so Tripeptide has two peptide bonds.
8. Which of the following options contains correct match of the following

| A. Antifertility Drugs | P. Arsphenamine |
| :--- | :--- |
| B. Antibiotics | Q. Norethindrone |
| C. Tranquilizers | R. Seldane |
| D. Antihistamines | S. Meprobamate |

A. $A-Q, B-P, C-S, D-R$
B. $A-P, B-Q, C-R, D-S$
C. $A-S, B-R, C-Q, D-P$
D. $A-S, B-R, C-P, D-Q$

## Answer (B)

Sol.
Antifertility Drugs - Norethindrone
Antibiotics - Arsphenamine
Tranquilizers - Meprobamate
Antihistamines - Seldane
9. When NaOH is added slowly to Benzoic acid, then which of the following plot of conductance Vs amount of NaOH will be correct
A.

B.

C.

D.

.

## Answer (B)

Sol.
First there are only Benzoic acid is present Which is weak acid
So, $\mathrm{PhCOOH} \leftrightarrow \mathrm{PhCOO}^{-}+\mathrm{H}^{+}$
From Beginning the Conductance increases Slightly
After adding NaOH in that it is make PhCOONa which is a salt and the conductance increases
When all $\mathrm{Na}^{+}$reacts with $\mathrm{PhCOO}^{-}$then $\mathrm{Only}^{\mathrm{OH}}{ }^{-}$Present in the solution and the conductance increases very highly

So, option B is the correct answer.
$=\left(T_{3}>T_{2}>T_{1}\right)$
And the correct option is (D)
7. The number of Peptide bonds present in Tripeptide VAL - PRO - GLY is
A. 1
B. 2
C. 3
D. 4

## Answer (B)

Sol.
In between two Amino Acids there are 1 peptide bond in question 3 Amino acids are given so Tripeptide has two peptide bonds.
8. Which of the following options contains correct match of the following

| A. Antifertility Drugs | P. Arsphenamine |
| :--- | :--- |
| B. Antibiotics | Q. Norethindrone |
| C. Tranquilizers | R. Seldane |
| D. Antihistamines | S. Meprobamate |

A. $A-Q, B-P, C-S, D-R$
B. $A-P, B-Q, C-R, D-S$
C. $A-S, B-R, C-Q, D-P$
D. $A-S, B-R, C-P, D-Q$

## Answer (B)

Sol.
Antifertility Drugs - Norethindrone
Antibiotics - Arsphenamine
Tranquilizers - Meprobamate
Antihistamines - Seldane
9. When NaOH is added slowly to Benzoic acid, then which of the following plot of conductance Vs amount of NaOH will be correct
12. Find the number of mono chloro products (structural only) which are obtained in the following reaction is


## Answer (5)

Sol.



13. Which of the following Lanthanoid ions is the best oxidising agent?
A. $L u^{2+}$
B. $C e^{2+}$
C. $C e^{4+}$
D. $\mathrm{Sn}^{2+}$

## Answer (C)

## Sol.

The most stable oxidation state of lanthanoids is +3 . The $\mathrm{Lu}^{2+}, \mathrm{Ce}^{2+}$ and $\mathrm{Sn}^{2+}$ will function as reducing agent because they will easily get oxidised to +3 oxidation state in each case. But $C e^{4+}$ will function as as oxidising agent and get reduced to $\mathrm{Ce}^{3+}$
$C e^{4+}+e^{-} \rightarrow \mathrm{Ce}^{3+}$
14. The number of ' $s$ ' electrons in unipositive state of an element having 55 protons in its nucleus is
A. 10
B. 8
C. 11
D. 12

## Answer (A)

## Sol.

The element is Cs. Cs+ has 10 ' $s$ ' electrons in its nucleus
15. Select the correct statement about physisorption
A. Physisorption is highly specific
B. Physisorption is always monolayer
C. Physisorption doesn't require activation energy
D. Physisorption is associated with very high enthalpy of adsorption

## Answer (C)

Sol.
Physisorption doesn't require activation energy
16. How many s-electrons are there in a Br -atom (Atomic No : of $\mathrm{Br}=35$ )

## Answer (8)

Sol.
Electronic configuration of Br atom is $1 s^{2} 2 s^{2} 2 p^{6} 3 s^{2} 3 p^{6} 3 d^{10} 4 s^{2} 4 p^{5}$
Therefore No of $s$ - electrons in a Br atom is 8 .
17.


I $\rightarrow$ II : Isobaric
II $\rightarrow$ III : Isochoric
III $\rightarrow$ I : Isothermal
All process are reversible
Find out the work done by the Gas for complete cyclic process (In atm.Lit)
(Report your answer to closest integer)

Sol.

$$
\begin{aligned}
& \mathrm{W} \rightarrow \rightarrow \|=-P \Delta V=-1 \mathrm{X}(40-20)=-20 \text { atm.it } \\
& \mathrm{W} \| \rightarrow{ }_{\| I}=0 \\
& \mathrm{~W}_{\mathrm{III}} \rightarrow \text { I }=2.303 n R T \cdot \log \frac{V 2}{V 1}=2.303 P V \log \frac{V 2}{V 1}=2.303(1 \mathrm{X} 20) \mathrm{Xlog} 22 \\
& =+13.818 \\
& \mathrm{~W} \rightarrow \| \mapsto \mathrm{II} \rightarrow \mathrm{IV}=-20+13.818 \\
& =-6.182 \text { atm. lit }
\end{aligned}
$$

Work done by the gas $=+6.182$ atm. lit
18. Find the sum of number of unpaired electrons in the following diatomic molecules: $N_{2}, N_{2}^{+}, \mathrm{O}_{2}, \mathrm{O}_{2}^{+}$?

## Answer (4.00)

## Sol.

No. of unpaired electrons in $N_{2}=0$
No. of unpaired electrons in $\mathrm{N}_{2}^{+}=1$
No. of unpaired electrons in $\mathrm{O}_{2}=2$
No. of unpaired electrons in $O_{2}^{+}=1$
Sum $=0+1+2+1=4$
19. $\mathrm{pK}_{\mathrm{a}}$ of lactic acid is 4 , Find the pH of 0.005 M calcium lactate at $27^{\circ} \mathrm{C}$ is?

## Answer (8)

Sol.

$$
\begin{aligned}
& p^{H}= \\
& \frac{1}{2}\left(p K_{w}+p K_{a}+\log C\right) \\
&= \frac{1}{2}\left(14+4+\log (0.01)=\frac{1}{2}(18-2)=\frac{1}{2}(16)=8\right.
\end{aligned}
$$

20. Find the sum of number of $\pi$ bonds in peroxydisulphuric acid and pyrosulphuric acid ?

## Answer (8)

## Sol.

Peroxydisulphuric acid $\left(\mathrm{H}_{2} \mathrm{~S}_{2} \mathrm{O}_{8}\right)$ has $4 \pi$ bonds
Pyrodisulphuric $\left(\mathrm{H}_{2} \mathrm{~S}_{2} \mathrm{O}_{7}\right)$ acid has $4 \pi$ bonds
Total number of $\pi$ bonds $=8$
21. How many of the following concentration terms are temperature independent?

Mole fraction,
Mass percent (\% w/w),
Molarity (M)

Molality (m)
ppm
volume percent (\%V/V)

## Answer (4:00)

Sol.
Temperature independent concentration terms are :
Mole fraction
Molality (m)
Parts Per Million (ppm)
Mass percentage (\%w/w)
22. One atom of $X$ has 25 MeV energy. The energy in 102 g of X is $\mathrm{P} \times 10^{25} \mathrm{MeV}$. Then find the value of ' $P$ '? Given : X has molar mass $=61 \mathrm{~g}$ and $\mathrm{N}_{\mathrm{A}}=6 \times 10^{23}$

## Answer (3)

Sol.

$$
\text { Total energy }=25 \times \frac{102}{61} \times 6 \times 10^{23}=3 \times 10^{25} \mathrm{MeV}
$$

23. If the ratio $\frac{c_{p}}{c_{v}}$ for monoatomic gases is $r_{1}$ and similar ratio for diatomic gas is $r_{2}$. Then the value of $\frac{r_{1}}{r_{2}} \times 21$ is

Answer (25)

Sol.

$$
\begin{aligned}
& \mathrm{r}_{1}=\frac{5}{3} \\
& \mathrm{r}_{2}=\frac{7}{5} \\
& \frac{r_{1}}{r_{2}}=\frac{25}{21} \\
& \frac{r_{1}}{r_{2}} \times 21=25
\end{aligned}
$$

## MATHEMATICS

1. $\int_{\frac{3 \sqrt{2}}{4}}^{\frac{3 \sqrt{3}}{4}} \frac{48}{\sqrt{9-4 x^{2}}} d x$ is equal to:
A. $\pi$
B. $2 \pi$
C. $-2 \pi$
D. $3 \pi$

## Answer (B)

Sol.

$$
\begin{aligned}
& I=\int_{\frac{3 \sqrt{2}}{4}}^{\frac{3 \sqrt{3}}{4}} \frac{48}{3 \times \sqrt{1-\frac{4 x^{2}}{9}}} d x \\
& \text { Let } \frac{2 x}{3}=t \\
& I=16 \int_{\frac{\sqrt{2}}{2}}^{\frac{\sqrt{3}}{2}} \frac{\frac{3}{2}}{\sqrt{1-t^{2}}} d t \\
& I=\left.24 \sin ^{-1} t\right|_{\frac{\sqrt{2}}{2}} ^{\frac{\sqrt{3}}{2}} \\
& I=24\left(\frac{\pi}{3}-\frac{\pi}{4}\right)=2 \pi
\end{aligned}
$$

2. $\left(\frac{1+\cos \left(\frac{2 \pi}{9}\right)+i \sin \left(\frac{2 \pi}{9}\right)}{1+\cos \left(\frac{2 \pi}{9}\right)-i \sin \left(\frac{2 \pi}{9}\right)}\right)^{3}$ is equal to:
A. $-\frac{1}{2}+\frac{i \sqrt{3}}{2}$
B. $-\frac{1}{2}-\frac{i \sqrt{3}}{2}$
C. $\frac{1}{2}-\frac{i \sqrt{3}}{2}$
D. $\frac{1}{2}+\frac{i \sqrt{3}}{2}$

## Answer (A)

Sol.

$$
\begin{aligned}
& \left(\frac{1+\cos \left(\frac{2 \pi}{9}\right)+i \sin \left(\frac{2 \pi}{9}\right)}{1+\cos \left(\frac{2 \pi}{9}\right)-i \sin \left(\frac{2 \pi}{9}\right)}\right)^{3}=\left(\frac{2 \cos \left(\frac{\pi}{9}\right)\left(\cos \frac{\pi}{9}+i \sin \left(\frac{\pi}{9}\right)\right)}{2 \cos \left(\frac{\pi}{9}\right)\left(\cos \frac{\pi}{9}-i \sin \left(\frac{\pi}{9}\right)\right)}\right)^{3} \\
& =\left(e^{\frac{i 2 \pi}{9}}\right)^{3} \\
& =e^{\frac{i 2 \pi}{3}} \\
& =-\frac{1}{2}+\frac{i \sqrt{3}}{2}
\end{aligned}
$$

3. If $\vec{a}=\hat{\imath}+2 \hat{\jmath}+m \hat{k}$ and $\vec{b}=\hat{\imath}-2 \hat{\jmath}+m \hat{k}$ if $\vec{a} \& \vec{b}$ are perpendicular to each other then $m$ equals:
A. $\pm \sqrt{2}$
B. $\pm \sqrt{3}$
C. $\pm 2$
D. $\pm \sqrt{5}$

## Answer (B)

## Sol.

If $\vec{a} \& \vec{b}$ are perpendicular to each other then $\vec{a} \cdot \vec{b}=0$
$\Rightarrow(\hat{\imath}+2 \hat{\jmath}+m \hat{k}) \cdot(\hat{\imath}-2 \hat{\jmath}+m \hat{k})=0$
$\Rightarrow 1-4+m^{2}=0$
$\Rightarrow m^{2}=3$
$\Rightarrow m= \pm \sqrt{3}$
4. The sum of the coefficients of first three terms in the expansion of $\left(x-\frac{3}{x^{2}}\right)^{n}$ is 376 . The coefficient of $x^{4}$ is equal to:
A. 695
B. 410
C. 405
D. 395

## Answer (C)

Sol.
The first three of $\left(x-\frac{3}{x^{2}}\right)^{n}={ }^{n} C_{0} x^{n}-{ }^{n} C_{1} \cdot 3 x^{n-3}+{ }^{n} C_{2} \cdot 9 x^{n-6}$
${ }^{n} C_{0}-{ }^{n} C_{1} \cdot 3+{ }^{n} C_{2} \cdot 9=376$
$\Rightarrow 1-3 n+\frac{n(n-1)}{2} \times 9=376$
$\Rightarrow 3 n^{2}-5 n-250=0$
$\Rightarrow n=10$
Now,
$T_{r+1}={ }^{10} C_{r} x^{10-r} \cdot\left(-\frac{3}{x^{2}}\right)^{r}$
$\Rightarrow{ }^{10} C_{r}(-3)^{r} \cdot x^{10-3 r}$
$\Rightarrow 10-r-2 r=4$
$\Rightarrow r=2$
Coefficient of $x^{4}={ }^{10} C_{2}(-3)^{2}$
$=45 \times 9=405$
5. Let $A=\{a, b, c, d\}$ and a relation $A \rightarrow A$ be $R=\{(a, b),(b, d),(b, c),(b, a)\}$ then minimum number of elements required to make $R$ equivalent is:
A. 7
B. 10
C. 12
D. 14

## Answer (C)

Sol.
Adding $(a, a),(b, b),(c, c),(d, d)$ makes it reflexive.
Adding ( $d, b$ ) and $(c, b)$ makes it symmetric.
And adding $(a, d),(a, c)$ makes it transitive.
So further $(d, a) \&(c, a)$ to be added to maintain symmetricity of relation.
further $(c, d) \&(d, c)$ also be added.
Hence total of 12 elements to be added to mole equivalence.
6. 3 urns $A, B, C$ contain 4 red, 6 black; 5 red, 5 black; $\lambda$ red, 4 black balls. A ball is drawn and found to be red. If probability that ball was drawn from urn $C$ is 0.4 , then the square of side of equilateral triangle inscribed in parabola $y^{2}=\lambda x$ with one vertex at vertex of parabola is
A. 144
B. 432
C. 368
D. 284

## Answer (B)

Sol.
$P($ Red ball from urn $C)=\frac{\frac{1}{3} \cdot \lambda}{\frac{1}{3} \cdot \frac{1}{10}+\frac{1}{3} \cdot \frac{5}{10}+\frac{1}{3} \cdot \frac{\lambda}{\lambda+4}}=\frac{4}{10}$
$\Rightarrow \lambda=6$
$m_{o p}=\frac{1}{\sqrt{3}}$
$\frac{2}{t}=\frac{1}{\sqrt{3}}$
Length of side $=4 a t=12 \sqrt{3}$ units
Square of side length $=432$

7. Total number of numbers formed using digits $3,5,6,7,8$ (without repetition) which are greater than 7000 , is equal to:
A. 148
B. 168
C. 144
D. 124

## Answer (B)

Sol.
Number using all the 5 digits $=5!=120$
Number using 4 digits
Case I:
When 7 is fixed at 1000 's place
7 $\qquad$ $=24$ ways
Case II:
When 8 is fixed at 1000 's place

8 $\qquad$ $=24$ ways
Total number $=120+24+24$
$=168$
8. A $5 \times 5$ matrix whose each entry is either 0 or 1 , is such that sum of entries of each column as well as each row is 1 . Number of such matrices is :
A. 30
B. 60
C. 90
D. 120

## Answer (D)

Sol.
In first column, 1 can be placed in any of the 5 places $=5$
In second column, 1 can be placed in any of the 4 places $=4$
In third column, 1 can be placed in any of the 3 places $=3$
In fourth column, 1 can be placed in any of the 2 places $=2$
In fifth column, 1 can be placed in only 1 place $=1$
Total $=5 \times 4 \times 3 \times 2 \times 1=120$
9. If the shortest distance between the lines $\frac{x-\sqrt{6}}{1}=\frac{y+\sqrt{6}}{2}=\frac{z-\sqrt{6}}{3}$ and $\frac{x-\lambda}{3}=\frac{y-2 \sqrt{6}}{4}=\frac{z-3 \sqrt{6}}{5}$ is 6 , then the sum of squares of all possible values of $\lambda$ is equal to:
A. 1024
B. 732
C. 416
D. 312

## Answer (B)

Sol.
$\overrightarrow{b_{1}}=\hat{\imath}+2 \hat{\jmath}+3 \hat{k}$
$\overrightarrow{b_{2}}=3 \hat{\imath}+4 \hat{\jmath}+5 \hat{k}$
$\overrightarrow{a_{2}}-\overrightarrow{a_{1}}=(\lambda-\sqrt{6}) \hat{\imath}+3 \sqrt{6} \hat{\jmath}+2 \sqrt{6} \hat{k}$
$\overrightarrow{b_{1}} \times \overrightarrow{b_{2}}=-2 \hat{\imath}+4 \hat{\jmath}-2 \hat{k}$
$d=\left|\frac{\left(\overrightarrow{a_{2}}-\overrightarrow{-a}\right) \cdot\left(\overrightarrow{b_{1}} \times \overrightarrow{b_{2}}\right)}{\left|\overrightarrow{b_{1}} \times \overrightarrow{b_{2}}\right|}\right| \Rightarrow 6=\left|\frac{-2(\lambda-\sqrt{6})+12 \sqrt{6}-4 \sqrt{6}}{\sqrt{24}}\right|$
$\Rightarrow \lambda=11 \sqrt{6},-\sqrt{6}$
$\Rightarrow \lambda_{1}^{2}+\lambda_{2}^{2}=732$
10. The proposition $\sim(p \wedge(p \rightarrow \sim q))$ is equivalent to :
A. $p \wedge(p \vee q)$
B. $\sim p \vee q$
C. $p \vee q$
D. $\sim p \wedge q$

## Answer (B)

Sol.

$$
\begin{aligned}
& \sim(p \wedge(p \rightarrow \sim q)) \\
& =\sim(p \wedge(\sim p \vee \sim q)) \\
& =\sim(f \vee(p \wedge \sim q)) \\
& =\sim(p \wedge \sim q) \\
& =\sim p \vee q
\end{aligned}
$$

11. $\left({ }^{30} C_{1}\right)^{2}+2 \cdot\left({ }^{30} C_{2}\right)^{2}+\cdots+30 \cdot\left({ }^{30} C_{30}\right)^{2}=\frac{\alpha \cdot 60!}{(30!)^{2}}-1$ then $\alpha$ is:
A. 12
B. 15
C. 10
D. 13

## Answer (B)

Sol.

$$
\begin{aligned}
& \sum_{r=1}^{30} r \cdot{ }^{30} C_{r} \cdot{ }^{30} C_{r}=30 \sum_{r=1}^{30}{ }^{29} C_{r-1} \cdot{ }^{30} C_{30-r} \\
& =30 \cdot \text { coefficient of } x^{29} \text { in }(1+x)^{29} \cdot(1+x)^{30} \\
& =30 \cdot{ }^{59} C_{29} \\
& =30 \cdot \frac{59!}{29!30!} \\
& \Rightarrow \frac{900 \cdot 59!}{(30!)^{2}} \\
& \Rightarrow \frac{15 \cdot 60!}{(30!)^{2}} \\
& \Rightarrow \alpha=15
\end{aligned}
$$

12. If $\lim _{x \rightarrow \alpha}|[x-5]-[2 x+2]|=0$ then:
A. $\alpha \in(-7.5,-6.5)$
B. $\alpha \in[-7.5,-6.5)$
C. $\alpha \in(-7.5,-6.5]$
D. $\alpha \in[-7.5,-6.5]$

## Answer (B)

Sol.

$$
\begin{aligned}
& \lim _{x \rightarrow \alpha}|[x-5]-[2 x+2]| \\
& \lim _{x \rightarrow \alpha}|[x]-5-[2 x]-2| \\
& \Rightarrow \lim _{x \rightarrow \alpha}|[x]-[2 x]-7| \\
& \lim _{x \rightarrow-7.5^{+}}|[x]-[2 x]-7| \\
& \Rightarrow|-8+15-7|=0
\end{aligned}
$$

$$
\begin{aligned}
& \text { At } x=-7.5,|-8+15-7|=0 \\
& \lim _{x \rightarrow-6.5^{-}}|[x]-[2 x]-7| \\
& \Rightarrow|-7+14-7|=0 \\
& \text { At } x=-6.5,|[x]-[2 x]-7| \\
& \Rightarrow|-7+13-7| \neq 0 \\
& \therefore \alpha \in[-7.5,-6.5)
\end{aligned}
$$

13. The locus of the mid points of chords of the circle $(x-4)^{2}+(y-5)^{2}=4$ which subtends angle $\theta_{i}$ at the centre of this circle is a circle of radius $r_{i}$. If $\theta_{1}=\frac{\pi}{3}, \theta_{2}=\frac{2 \pi}{3}$ and $r_{1}^{2}=r_{2}^{2}+r_{3}^{2}$ then $\theta_{3}$ is equal to:
A. $\frac{\pi}{2}$
B. $\frac{\pi}{12}$
C. $\frac{\pi}{6}$
D. $\frac{\pi}{4}$

## Answer (A)

Sol.
If a chord of circle of radius $R$ subtends angle $\theta_{i}$ at the centre then locus of the midpoint of this chord is a circle of radius $r_{i}=R \cdot \cos \left(\frac{\theta_{i}}{2}\right)$
Given,

$$
\begin{aligned}
& r_{1}^{2}=r_{2}^{2}+r_{3}^{2} \\
& \Rightarrow \cos ^{2} \frac{\theta_{1}}{2}=\cos ^{2} \frac{\theta_{2}}{2}+\cos ^{2} \frac{\theta_{3}}{2} \\
& \Rightarrow \cos ^{2} \frac{\pi}{6}=\cos ^{2} \frac{\pi}{3}+\cos ^{2} \frac{\theta_{3}}{2} \\
& \Rightarrow \frac{3}{4}=\frac{1}{4}+\cos ^{2} \frac{\theta_{3}}{2} \\
& \Rightarrow \cos ^{2} \frac{\theta_{3}}{2}=\frac{1}{2} \\
& \Rightarrow \cos \frac{\theta_{3}}{2}=\frac{1}{\sqrt{2}} \\
& \Rightarrow \frac{\theta_{3}}{2}=\frac{\pi}{4} \\
& \therefore \theta_{3}=\frac{\pi}{2}
\end{aligned}
$$

14. Let $f(x)=x^{3}-x^{2} f^{\prime}(1)+x f^{\prime \prime}(2)-f^{\prime \prime \prime}(3)$, then:
A. $\quad f(0)=f(1)+f(2)+f(3)$
B. $f(3)+2 f(0)=f(2)+f(1)$
C. $2 f(0)=f(1)-f(2)$
D. $f(3)-f(1)=2 f(2)$

## Answer (B)

Sol.

$$
\begin{aligned}
& \text { Let } f(x)=x^{3}-A x^{2}+B x-C \\
& \Rightarrow f^{\prime}(1)=3-2 A+B=A
\end{aligned}
$$

$$
\begin{aligned}
& \Rightarrow f^{\prime \prime}(2)=12-2 A=B \\
& \Rightarrow f^{\prime \prime \prime}(3)=6=C
\end{aligned}
$$

Solving, $f(x)=x^{3}-3 x^{2}+6 x-6$
$f(0)=-6, f(1)=-2$
$f(2)=8-12+12-6=2$
$f(3)=27-27+18-6=12$
$\therefore f(3)+2 f(0)=f(2)+f(1)$
15. $\frac{1^{3}+2^{3}+3^{3}+\cdots \text { up to } n \text { terms }}{1 \cdot 3+2 \cdot 5+3 \cdot 7+\cdots+\text { up to } n \text { terms }}=\frac{9}{5}$ then $n$ is equal to:
A. 5
B. 7
C. 12
D. 9

## Answer (B)

Sol.

$$
\begin{aligned}
& \frac{\mathbf{1}^{3}+2^{3}+\mathbf{3}^{3}+\cdots \text { up to } n \text { terms }}{\mathbf{1} \cdot \mathbf{3}+\mathbf{2} \cdot \mathbf{5}+\mathbf{3} \cdot \mathbf{7}+\cdots+\text { up to } n \text { terms }}= \\
& \frac{\mathbf{1}^{3}+\mathbf{2}^{3}+\mathbf{3}^{3}+\cdots+\mathbf{n}^{3}}{\sum \boldsymbol{n}(\mathbf{2 n + 1})} \\
& \Rightarrow \frac{\frac{n^{2}(n+1)^{2}}{4}}{2 \cdot \frac{n(n+1)(2 n+1)}{6}+\frac{n(n+1)}{2}}=\frac{9}{5} \\
& \Rightarrow \frac{\frac{n(n+1)}{4} \cdot 6}{2(2 n+1)+3}=\frac{9}{5} \\
& \Rightarrow 5 n^{2}-19 n-30=0 \\
& \Rightarrow 5 n^{2}-25 n+6 n-30=0 \\
& \quad \Rightarrow n=5,-\frac{6}{5} \Rightarrow n=5
\end{aligned}
$$

16. If $|\operatorname{adj}(\operatorname{adj}(\operatorname{adj} A))|=12^{4}$ then $\left|\boldsymbol{A}^{-1}(\operatorname{adj} A)\right|$ equals:( where $\boldsymbol{A}$ is matrix of order $\mathbf{3} \times 3$ )
A. $2 \sqrt{3}$
B. 1
C. 6
D. 12

## Answer (A)

Sol.

$$
\begin{aligned}
& |\operatorname{adj}(\operatorname{adj}(\operatorname{adj} A))|=|A|^{2^{3}}=|A|^{8}=12^{4} \\
& \therefore|A|=2 \sqrt{3} \\
& \left|A^{-1} \operatorname{adj}(A)\right|=\left|A^{-1}\right| \cdot|\operatorname{adj}(A)|=\frac{1}{|A|} \cdot|A|^{2}=|A|=2 \sqrt{3}
\end{aligned}
$$

17. If area bound between $y^{2}-4 y=-x$ and $x+y=0$ is $A$ then $6 A$ equals.

## Answer (125/6)

Sol.


Point of intersection of $y^{2}-4 y=-x$ and $x+y=0$ is

$$
\begin{aligned}
& y^{2}-4 y=y \\
& \Rightarrow y^{2}-5 y=0 \\
& \Rightarrow y=5 \text { or } y=0 \\
& \therefore x=-5 \text { or } x=0
\end{aligned}
$$

Required Area $=\int_{0}^{5}\left(4 y-y^{2}\right)-(-y) d y$
$=\int_{0}^{5}\left(5 y-y^{2}\right) d y$
$=\left[\frac{5 y^{2}}{2}-\frac{y^{3}}{3}\right]_{0}^{5}$
$\Rightarrow \frac{125}{2}-\frac{125}{3}=\frac{125}{6}$
18. Let $a_{1}, a_{2}, \ldots, a_{6}$ be in A.P. such that $a_{1}+a_{3}=10 \&$ mean of $a_{1}, a_{2}, \ldots, a_{6}$ is $\frac{19}{2}$. Then $8 \sigma^{2}$ is equal to $\qquad$ -

Answer (210)

Sol.
$a_{1}+a_{3}=10$
$\Rightarrow 2 a+2 d=10$

$$
\Rightarrow a+d=5
$$

Also, $\frac{a+(a+d)+\cdots+(a+5 d)}{6}=\frac{19}{2}$
$\Rightarrow 2 a+5 d=19$
$\therefore a=2 \& d=3$
$\therefore$ Given A.P. is $2,5,8,11,14,17$

$$
\Rightarrow \sigma^{2}=\frac{\left(2-\frac{19}{2}\right)^{2}+\left(5-\frac{19}{2}\right)^{2}+\cdots+\left(17-\frac{19}{2}\right)^{2}}{6}=26.25
$$

$$
\therefore 8 \sigma^{2}=8 \times 26.25=210
$$

